#### Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

### (1*S*,2*S*,3*R*,4*S*,5*R*,7*S*,8*S*,10*R*,13*S*)-2-Debenzoyl-10-deacetyl-2-(3-fluorobenzoyl)-7,10-bis(2,2,2-trichloroethoxycarbonyl)baccatin III ethyl acetate monosolvate monohydrate

#### Chen Zhang, Cheng Xie, Jun Chang, Hong-Fu Lu and Xun Sun\*

School of Pharmacy, Fudan University, 826 Zhangheng Road, Shanghai 201203, People's Republic of China Correspondence e-mail: sunxunf@shmu.edu.cn

Received 11 November 2010; accepted 20 January 2011

Key indicators: single-crystal X-ray study; T = 293 K; mean  $\sigma$ (C–C) = 0.006 Å; disorder in main residue; *R* factor = 0.051; *wR* factor = 0.115; data-to-parameter ratio = 14.5.

In the title compound,  $C_{35}H_{37}Cl_6FO_{14}\cdot C_4H_8O_2\cdot H_2O$ , the absolute configurations (1S,2S,3R,4S,5R,7S,8S,10R,13S) for the nine chiral centres of the molecule has been determined. In the crystal, molecules are linked by  $O-H\cdots O$  and  $O-H\cdots Cl$  hydrogen bonds.

#### **Related literature**

For the preparation of the title compound, an intermediate from the synthesis of a fluorinated docetaxel analog, see: Lu *et al.* (2009). For the absolute configuration of the title compound, see: Kingston *et al.* (1982).



#### Experimental

Crystal data C<sub>35</sub>H<sub>37</sub>Cl<sub>6</sub>FO<sub>14</sub>·C<sub>4</sub>H<sub>8</sub>O<sub>2</sub>·H<sub>2</sub>O

 $M_r = 1019.47$ 

Orthorhombic,  $P_{2_12_12_1}$  a = 14.7037 (11) Å b = 16.6601 (12) Å c = 18.9258 (14) Å $V = 4636.2 (6) \text{ Å}^3$ 

#### Data collection

Bruker SMART CCD area-detector diffractometer Absorption correction: multi-scan (*SADABS*; Bruker, 2001)  $T_{\rm min} = 0.604, T_{\rm max} = 1.000$ 

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.051$   $wR(F^2) = 0.115$  S = 0.919094 reflections 628 parameters 101 restraints Z = 4Mo K\alpha radiation  $\mu = 0.44 \text{ mm}^{-1}$ T = 293 K $0.40 \times 0.31 \times 0.29 \text{ mm}$ 

25522 measured reflections 9094 independent reflections 6096 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.053$ 

H atoms treated by a mixture of independent and constrained refinement 
$$\begin{split} &\Delta\rho_{max}=0.64\ e\ \mathring{A}^{-3}\\ &\Delta\rho_{min}=-0.32\ e\ \mathring{A}^{-3}\\ &Absolute structure: Flack (1983),\\ &4052\ Friedel pairs\\ &Flack parameter: 0.03 (6) \end{split}$$

Table 1			
Hydrogen-bond geometry	(Å,	°).	

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots \mathbf{A}$
$017 - H17E \cdots Cl1^{i}$ $017 - H17E \cdots O9^{i}$ $014 - H14 \cdots O17$	0.89(2) 0.89(2) 0.83(2)	2.85 (6) 2.48 (7) 2.00 (2)	3.582 (5) 3.241 (6) 2.791 (6)	141 (6) 143 (9) 161 (4)
$01 - H1 \cdot \cdot \cdot O15$ $017 - H17D \cdot \cdot \cdot O1^{ii}$	$\begin{array}{c} 0.03 \ (2) \\ 0.82 \ (2) \\ 0.91 \ (2) \end{array}$	2.00 (2) 2.01 (3) 2.25 (4)	2.786 (6) 3.085 (6)	158 (5) 152 (6)

Symmetry codes: (i) -x + 1,  $y + \frac{1}{2}$ ,  $-z + \frac{3}{2}$ ; (ii)  $x + \frac{1}{2}$ ,  $-y + \frac{1}{2}$ , -z + 1.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

The work was supported financially by the National Natural Science Foundation of China (No. 20772017), the National Drug Innovative Program (No. 2009ZX09301–011) and the Shanghai Municipal Committee of Science and Technology (No. 07DZ19713). We would like to thank Dr Jie Sun for the single-crystal X-ray determination.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: ZS2080).

#### References

- Bruker (2001). SAINT, SMART and SADABS. Bruker AXS Inc., Madison, Wisconsin, USA.
- Flack, H. D. (1983). Acta Cryst. A39, 876-881.
- Kingston, D. G. I., Hawkins, D. R. & Ovington, L. (1982). J. Nat. Prod. 45, 466– 470.
- Lu, H.-F., Sun, X., Xu, L., Lou, L.-G. & Lin, G.-Q. (2009). Eur. J. Med. Chem. 44, 482–491.
- Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122.

Acta Cryst. (2011). E67, o547 [doi:10.1107/S1600536811002790]

# (1*S*,2*S*,3*R*,4*S*,5*R*,7*S*,8*S*,10*R*,13*S*)-2-Debenzoyl-10-deacetyl-2-(3-fluorobenzoyl)-7,10-bis(2,2,2-trichloroethoxycarbonyl)baccatin III ethyl acetate monosolvate monohydrate

#### C. Zhang, C. Xie, J. Chang, H.-F. Lu and X. Sun

#### Comment

In our research on the synthesis of a series of novel fluorinated docetaxel analogs, one of the key intermediate products, the title compound 7,10-di- (2,2,2-trichloroethyloxycarbonyl)-2-debenzoyl- 2-(3-fluorobenzoyl)-10-deacetyl-baccatin III monohydrate ethyl acetate monosolvate,  $C_{35}H_{37}Cl_6FO_{14}$ .  $C_4H_8O_2$ .  $H_2O$  (I) (Fig. 1) was obtained from 10-deacetylbaccatin III (Kingston *et al.*,1982). The reaction scheme (Lu *et al.*, 2009) is shown in Fig.2. Absolute configuration(1*S*,2*S*,3*R*,4*S*,5*R*,7*S*,8*S*,10*R*,13*S*) for the nine chiral centres of the molecule has been determined. In the crystal structure, molecules are linked by hydroxy and water O—H···O hydrogen bonds (Table 1).

#### **Experimental**

To a solution of 2-debenzoyl-2-(*m*-fluorobenzoyl)-10-deacetylbaccatin III (1.03 g, 1.84 mmol) in anhydrous pyridine (20 ml) was added 2,2,2-trichloroethylchloroformate (0.85 ml, 6.17 mmol) dropwise at 273 K. The reaction mixture was then warmed to room temperature and further stirred for 30 min. The reaction mixture was then quenched with water and the solvent was removed under reduced pressure. The residue was dissolved in DCM, and washed with dilute HCl and brine. The organic layer was dried with anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated *in vacuo*. The crude residue was purified by flash column chromatography (petroleum ether/ ethyl acetate 2/1) to give the title compound (1.58 g, 94% yield) as a white solid. Suitable crystals were obtained by recrystallization from hexane and DCM (m.p. 495–497 K). <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): d 1.12 (s, 3H), 1.15 (s, 3H), 1.85 (s, 3H), 2.17 (s, 3H), 2.30 (m, 2H), 2.31 (s, 3H), 2.05 and 2.65 (2 m, 2H), 3.98 (d, 1H, J = 6.6 Hz), 4.14 and 4.33 (2 d, 2H, J = 8.4 Hz), 4.61 and 4.92 (2 d, 2H, J=12.0 Hz), 4.78 (d, 2H, J = 12.0 Hz), 4.90 (m, 1H), 5.00 (d, 1H, J = 7.8 Hz), 5.59 (m, 1H), 5.62 (d, 1H, J = 7.5 Hz), 6.27 (s, 1H), 7.33 (m, 1H), 7.48 (m, 1H), 7.79 (m, 1H), 7.90 (d, 1H, J = 7.5 Hz); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): d 10.6, 15.4, 20.1, 22.5, 26.6, 33.3, 38.4, 42.6, 47.4, 56.3, 67.8, 74.6, 76.2, 76.6, 77.1, 77.4, 78.7, 79.7, 80.4, 83.7, 94.2, 94.3, 116.9, 120.9, 125.9, 130.4, 130.8, 131.4, 146.6, 153.2, 153.3, 162.6 (d, J<sub>C—F</sub> = 246.4 Hz), 165.7, 170.8, 201.1; ESIMS m/z 933.0 [*M* + Na<sup>+</sup>]; HRMS (MALDI) m/z calcd for  $C_{35}H_{37}Cl_6FO_{14}Na^+$  [*M* + Na<sup>+</sup>]; 933.0215, found: 933.01908.

#### Refinement

Hydrogen atoms of the hydroxy groups and the water molecule were located by difference methods and both positional and isotropic displacement parameters were refined. Other H atoms were positioned geometrically and treated as riding with C—H = 0.96–0.98 Å and  $U_{iso}$  = 1.2 or 1.5Ueq(C). The F atom was significantly disordered and was subsequently refined isotropically. The absolute configuration for the nine chiral centres in the molecule have been assigned C1(S), C2(S), C3(*R*), C4(S), C5(*R*), C7(S), C8(S), C10(*R*), C13(S) on the basis of the Flack structure parameter [0.03 (6)] (Flack,1983) (atom mames are those arbitrarily assigned in this crystallographic study).

**Figures** 



Fig. 1. Molecular configuration and atom numbering scheme for (I) with probaility ellipsoids drawn at the 40% probability level???.

Fig. 2. Reaction scheme for the synthesis of (I)

## (1*S*,2*S*,3*R*,4*S*,5*R*,7*S*,8*S*, 10*R*,13*S*)-2-Debenzoyl-10-deacetyl-2-(3-fluorobenzoyl)-7,10- bis(2,2,2-trichloroethoxycarbonyl)baccatin III monohydrate ethyl acetate monosolvate

Crystal data

$C_{35}H_{37}Cl_{6}FO_{14}{\cdot}C_{4}H_{8}O_{2}{\cdot}H_{2}O$	$D_{\rm x} = 1.461 {\rm Mg m}^{-3}$
$M_r = 1019.47$	Melting point = 495–497 K
Orthorhombic, <i>P</i> 2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>	Mo <i>K</i> $\alpha$ radiation, $\lambda = 0.71073$ Å
Hall symbol: P 2ac 2ab	Cell parameters from 4411 reflections
a = 14.7037 (11)  Å	$\theta = 5.0 - 38.3^{\circ}$
b = 16.6601 (12)  Å	$\mu = 0.44 \text{ mm}^{-1}$
c = 18.9258 (14)  Å	T = 293  K
$V = 4636.2 (6) Å^3$	Prismatic, colorless
Z = 4	$0.40 \times 0.31 \times 0.29 \text{ mm}$
F(000) = 2112	

#### Data collection

Bruker SMART CCD area-detector diffractometer	9094 independent reflections
Radiation source: fine-focus sealed tube	6096 reflections with $I > 2\sigma(I)$
graphite	$R_{\rm int} = 0.053$
$\phi$ and $\omega$ scans	$\theta_{\text{max}} = 26.0^{\circ}, \ \theta_{\text{min}} = 1.6^{\circ}$
Absorption correction: multi-scan ( <i>SADABS</i> ; Bruker, 2001)	$h = -13 \rightarrow 18$
$T_{\min} = 0.604, \ T_{\max} = 1.000$	$k = -19 \rightarrow 20$
25522 measured reflections	$l = -22 \rightarrow 23$

Refinement

Refinement on $F^2$	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.051$	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2) = 0.115$	$w = 1/[\sigma^2(F_0^2) + (0.0513P)^2]$ where $P = (F_0^2 + 2F_c^2)/3$
<i>S</i> = 0.91	$(\Delta/\sigma)_{\rm max} = 0.013$
9094 reflections	$\Delta \rho_{max} = 0.64 \text{ e } \text{\AA}^{-3}$
628 parameters	$\Delta \rho_{\rm min} = -0.32 \text{ e } \text{\AA}^{-3}$
101 restraints	Absolute structure: Flack (1983), 4052 Friedel pairs
Primary atom site location: structure-invariant direct methods	Flack parameter: 0.03 (6)

#### Special details

**Geometry**. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement**. Refinement of  $F^2$  against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on  $F^2$ , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on  $F^2$  are statistically about twice as large as those based on *F*, and *R*-factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(A^2)$ 

	x	У	Ζ	$U_{\rm iso}$ */ $U_{\rm eq}$	Occ. (<1)
Cl1	0.32716 (9)	0.01873 (8)	1.04792 (7)	0.0699 (4)	
Cl2	0.48384 (11)	0.04791 (9)	1.13542 (6)	0.0841 (5)	
Cl3	0.45947 (9)	-0.10584 (7)	1.06832 (6)	0.0652 (4)	
Cl4	0.08093 (10)	0.25135 (9)	0.93518 (9)	0.0931 (5)	
C15	0.21962 (10)	0.36674 (7)	0.96872 (8)	0.0782 (5)	
C16	0.17098 (13)	0.24496 (9)	1.06934 (8)	0.1047 (6)	
01	0.3735 (2)	0.16138 (19)	0.49241 (15)	0.0514 (8)	
O2	0.48084 (18)	0.03348 (15)	0.50337 (12)	0.0380 (6)	
O3	0.3511 (2)	-0.02415 (19)	0.46551 (17)	0.0642 (9)	
O4	0.64110 (17)	0.00931 (15)	0.57227 (13)	0.0385 (6)	
O5	0.7005 (2)	0.05200 (18)	0.67458 (16)	0.0538 (8)	
O6	0.5595 (2)	-0.15701 (15)	0.62706 (14)	0.0488 (8)	
07	0.47088 (17)	-0.02112 (15)	0.82322 (12)	0.0365 (6)	
08	0.5838 (2)	0.0466 (2)	0.87704 (15)	0.0664 (10)	
09	0.46637 (18)	-0.00885 (15)	0.93540 (12)	0.0408 (7)	
O10	0.30922 (18)	0.05437 (17)	0.75999 (15)	0.0470 (7)	

011	0.34759 (18)	0.20512 (15)	0.78236 (12)	0.0378 (6)
012	0.3935 (2)	0.14650 (18)	0.88410 (14)	0.0578 (9)
O13	0.2698 (2)	0.22498 (16)	0.87740 (14)	0.0498 (8)
O14	0.6592 (2)	0.2433 (2)	0.59204 (18)	0.0644 (9)
O15	0.4525 (4)	0.2095 (4)	0.3655 (2)	0.163 (2)
O16	0.5049 (3)	0.1760 (3)	0.2632 (2)	0.1125 (16)
O17	0.7215 (3)	0.4008 (3)	0.6057 (3)	0.0948 (13)
C1	0.4306 (3)	0.1568 (2)	0.55338 (18)	0.0352 (9)
C2	0.4391 (3)	0.0639 (2)	0.56755 (18)	0.0338 (9)
H2	0.3779	0.0411	0.5718	0.041*
C3	0.4974 (2)	0.0350 (2)	0.63144 (17)	0.0275 (8)
H3	0.5330	0.0818	0.6464	0.033*
C4	0.5681 (3)	-0.0301 (2)	0.61001 (18)	0.0309 (8)
C5	0.6021 (3)	-0.0914 (2)	0.66434 (18)	0.0356 (9)
Н5	0.6684	-0.0962	0.6615	0.043*
C6	0.5727 (3)	-0.0846 (2)	0.74108 (19)	0.0396 (10)
H6A	0.6263	-0.0832	0.7709	0.047*
H6B	0.5377	-0.1318	0.7538	0.047*
C7	0.5157 (3)	-0.0099 (2)	0.75516 (17)	0.0312 (8)
H7	0.5562	0.0367	0.7583	0.037*
C8	0.4412 (2)	0.0076 (2)	0.69925 (18)	0.0290 (8)
C9	0.3801 (3)	0.0733 (2)	0.73402 (18)	0.0304 (8)
C10	0.4163 (3)	0.1582 (2)	0.74474 (18)	0.0314 (9)
H10	0.4702	0.1547	0.7750	0.038*
C11	0.4436 (3)	0.1997 (2)	0.67764 (18)	0.0325 (9)
C12	0.5294 (3)	0.2261 (2)	0.6700 (2)	0.0373 (9)
C13	0.5632 (3)	0.2488 (2)	0.5977 (2)	0.0455 (11)
H13	0.5451	0.3043	0.5881	0.055*
C14	0.5237 (3)	0.1953 (2)	0.54023 (19)	0.0460 (11)
H14A	0.5669	0.1525	0.5313	0.055*
H14B	0.5194	0.2268	0.4973	0.055*
C15	0.3795 (3)	0.2012 (2)	0.61382 (19)	0.0356 (9)
C16	0.3599 (3)	0.2890 (2)	0.5920 (2)	0.0498 (11)
H16A	0.3268	0.3154	0.6289	0.075*
H16B	0.3246	0.2895	0.5493	0.075*
H16C	0.4164	0.3165	0.5841	0.075*
C17	0.2842 (3)	0.1642 (2)	0.6257 (2)	0.0449 (10)
H17A	0.2906	0.1090	0.6395	0.067*
H17B	0.2499	0.1672	0.5826	0.067*
H17C	0.2532	0.1933	0.6622	0.067*
C18	0.5361 (3)	-0.1049 (2)	0.5696 (2)	0.0400 (10)
H18A	0.4715	-0.1045	0.5592	0.048*
H18B	0.5710	-0.1153	0.5271	0.048*
C19	0.3806 (3)	-0.0650 (2)	0.6853 (2)	0.0377 (9)
H19A	0.4176	-0.1099	0.6717	0.057*
H19B	0.3386	-0.0527	0.6480	0.057*
H19C	0.3474	-0.0781	0.7274	0.057*
C20	0.5979 (3)	0.2312 (3)	0.7288 (2)	0.0522 (12)
H20A	0.5668	0.2361	0.7732	0.078*

H20B	0.6361	0.2772	0.7217	0.078*	
H20C	0.6345	0.1835	0.7291	0.078*	
C21	0.4298 (4)	-0.0068(2)	0.4567 (2)	0.0463 (11)	
F1	0.4305 (8)	-0.1272 (5)	0.2284 (3)	0.198 (3)	0.70
C22	0.4730 (5)	-0.0350 (5)	0.3940 (3)	0.026 (2)	0.526 (16)
C23	0.4227 (6)	-0.0638 (5)	0.3372 (3)	0.041 (3)	0.526 (16)
H23	0.3595	-0.0655	0.3398	0.049*	0.526 (16)
C24	0.4667 (8)	-0.0900 (5)	0.2764 (3)	0.064 (3)	0.526 (16)
C25	0.5611 (8)	-0.0874 (5)	0.2725 (3)	0.083 (5)	0.526 (16)
H25	0.5906	-0.1049	0.2319	0.099*	0.526 (16)
C26	0.6114 (6)	-0.0587 (6)	0.3293 (4)	0.073 (4)	0.526 (16)
H26	0.6745	-0.0570	0.3267	0.087*	0.526 (16)
C27	0.5673 (5)	-0.0325 (5)	0.3901 (3)	0.042 (3)	0.526 (16)
H27	0.6010	-0.0133	0.4281	0.050*	0.526 (16)
F1'	0.5191 (10)	-0.1369 (6)	0.2353 (4)	0.197 (9)	0.30
C22'	0.5049 (9)	-0.0223 (7)	0.3965 (5)	0.061 (4)	0.474 (16)
C23'	0.4661 (9)	-0.0641 (7)	0.3405 (6)	0.081 (5)	0.474 (16)
H23'	0.4035	-0.0716	0.3388	0.097*	0.474 (16)
C24'	0.5210 (11)	-0.0946 (6)	0.2871 (5)	0.066 (4)	0.474 (16)
C25'	0.6146 (10)	-0.0835 (6)	0.2897 (5)	0.093 (5)	0.474 (16)
H25'	0.6513	-0.1039	0.2540	0.112*	0.474 (16)
C26'	0.6534 (8)	-0.0417 (7)	0.3457 (6)	0.112 (6)	0.474 (16)
H26'	0.7160	-0.0342	0.3475	0.135*	0.474 (16)
C27'	0.5985 (9)	-0.0112 (8)	0.3992 (5)	0.083 (5)	0.474 (16)
H27'	0.6244	0.0168	0.4366	0.100*	0.474 (16)
C28	0.7039 (3)	0.0475 (2)	0.6126 (3)	0.0469 (11)	
C29	0.7772 (3)	0.0835 (3)	0.5674 (3)	0.0630 (13)	
H29A	0.7614	0.1378	0.5557	0.094*	
H29B	0.7833	0.0526	0.5248	0.094*	
H29C	0.8338	0.0830	0.5927	0.094*	
C30	0.5148 (3)	0.0097 (2)	0.8779 (2)	0.0387 (9)	
C31	0.4976 (3)	0.0306 (3)	0.99812 (19)	0.0468 (11)	
H31A	0.4888	0.0881	0.9940	0.056*	
H31B	0.5619	0.0203	1.0050	0.056*	
C32	0.4442 (3)	-0.0013 (2)	1.0595 (2)	0.0462 (11)	
C33	0.3432 (3)	0.1873 (2)	0.8512 (2)	0.0440 (11)	
C34	0.2589 (4)	0.2100 (3)	0.9515 (2)	0.0605 (14)	
H34A	0.3157	0.2198	0.9761	0.073*	
H34B	0.2414	0.1545	0.9592	0.073*	
C35	0.1856 (4)	0.2657 (3)	0.9791 (2)	0.0627 (14)	
C36	0.3518 (5)	0.1838 (7)	0.2640 (5)	0.167 (4)	
H36A	0.3408	0.1283	0.2535	0.250*	
H36B	0.3562	0.2137	0.2208	0.250*	
H36C	0.3025	0.2045	0.2919	0.250*	
C37	0.4357 (7)	0.1913 (5)	0.3031 (4)	0.129 (3)	
C38	0.5947 (6)	0.1783 (5)	0.2986 (4)	0.137 (3)	
H38A	0.6089	0.2330	0.3124	0.164*	
H38B	0.5931	0.1455	0.3409	0.164*	
C39	0.6630 (5)	0.1490 (6)	0.2517 (5)	0.147 (3)	

H39A	0.6456	0.0970	0.2345	0.220*
H39B	0.7198	0.1449	0.2764	0.220*
H39C	0.6695	0.1853	0.2126	0.220*
H1	0.391 (4)	0.187 (3)	0.4583 (18)	0.08 (2)*
H14	0.689 (3)	0.2851 (17)	0.593 (2)	0.050 (14)*
H17D	0.749 (4)	0.376 (4)	0.569 (2)	0.13 (3)*
H17E	0.690 (7)	0.440 (5)	0.585 (4)	0.21 (7)*

### Atomic displacement parameters $(\text{\AA}^2)$

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Cl1	0.0532 (7)	0.0841 (9)	0.0724 (8)	0.0076 (7)	0.0175 (7)	0.0100 (7)
C12	0.1221 (13)	0.0941 (10)	0.0360 (6)	-0.0281 (9)	-0.0002 (7)	-0.0152 (6)
C13	0.0857 (10)	0.0550 (7)	0.0548 (7)	0.0017 (7)	0.0015 (7)	0.0104 (6)
Cl4	0.0709 (10)	0.0803 (10)	0.1283 (13)	-0.0074 (8)	0.0329 (10)	-0.0173 (10)
C15	0.0888 (10)	0.0376 (6)	0.1083 (11)	-0.0080 (7)	0.0518 (9)	-0.0079 (7)
C16	0.1549 (16)	0.0884 (11)	0.0707 (9)	-0.0151 (10)	0.0685 (10)	-0.0060 (8)
01	0.066 (2)	0.0535 (19)	0.0343 (17)	0.0101 (17)	-0.0127 (16)	0.0037 (15)
O2	0.0474 (17)	0.0383 (15)	0.0284 (13)	0.0033 (13)	0.0035 (13)	-0.0035 (11)
O3	0.065 (2)	0.061 (2)	0.066 (2)	-0.0042 (19)	-0.0187 (19)	-0.0175 (17)
O4	0.0352 (15)	0.0419 (15)	0.0383 (14)	0.0036 (13)	0.0098 (13)	0.0014 (13)
O5	0.0481 (19)	0.062 (2)	0.0513 (19)	-0.0068 (16)	-0.0071 (15)	-0.0079 (17)
O6	0.070 (2)	0.0274 (14)	0.0491 (16)	0.0024 (14)	-0.0006 (16)	0.0016 (13)
07	0.0407 (16)	0.0424 (15)	0.0265 (12)	-0.0062 (13)	0.0011 (12)	0.0014 (12)
08	0.052 (2)	0.108 (3)	0.0388 (16)	-0.0339 (19)	0.0029 (15)	-0.0104 (18)
09	0.0460 (17)	0.0527 (17)	0.0236 (12)	-0.0109 (14)	0.0009 (13)	0.0019 (12)
O10	0.0385 (17)	0.0447 (16)	0.0578 (18)	-0.0026 (14)	0.0187 (15)	-0.0051 (15)
011	0.0459 (17)	0.0351 (15)	0.0325 (14)	0.0119 (13)	0.0032 (13)	-0.0053 (12)
012	0.079 (2)	0.0545 (19)	0.0405 (16)	0.0324 (18)	0.0049 (16)	0.0056 (15)
O13	0.062 (2)	0.0432 (17)	0.0440 (16)	0.0147 (15)	0.0193 (16)	0.0019 (14)
O14	0.055 (2)	0.056 (2)	0.082 (2)	-0.0113 (18)	0.0204 (18)	-0.0080 (19)
015	0.196 (5)	0.232 (6)	0.061 (3)	-0.054 (5)	0.022 (3)	-0.007 (3)
O16	0.079 (3)	0.176 (5)	0.082 (3)	-0.019 (3)	0.013 (3)	-0.039 (3)
O17	0.072 (3)	0.077 (3)	0.136 (4)	-0.010 (2)	0.009 (3)	0.024 (3)
C1	0.045 (2)	0.034 (2)	0.0270 (19)	0.0055 (19)	-0.0045 (18)	0.0021 (16)
C2	0.035 (2)	0.034 (2)	0.0324 (19)	0.0007 (17)	0.0024 (18)	-0.0057 (17)
C3	0.030 (2)	0.0246 (19)	0.0283 (18)	-0.0001 (16)	0.0029 (16)	-0.0019 (15)
C4	0.038 (2)	0.027 (2)	0.0278 (18)	0.0017 (17)	0.0034 (17)	0.0009 (16)
C5	0.039 (2)	0.031 (2)	0.037 (2)	0.0076 (18)	0.0047 (19)	0.0002 (17)
C6	0.050 (3)	0.034 (2)	0.035 (2)	0.008 (2)	-0.001 (2)	0.0082 (18)
C7	0.036 (2)	0.033 (2)	0.0249 (17)	-0.0037 (17)	0.0041 (17)	0.0022 (16)
C8	0.030 (2)	0.0247 (19)	0.0321 (19)	-0.0005 (16)	0.0017 (16)	0.0025 (15)
C9	0.030 (2)	0.033 (2)	0.0282 (19)	0.0034 (17)	-0.0001 (17)	0.0001 (16)
C10	0.028 (2)	0.032 (2)	0.034 (2)	0.0094 (17)	0.0032 (17)	-0.0054 (17)
C11	0.041 (2)	0.0205 (19)	0.036 (2)	0.0052 (17)	0.0018 (19)	-0.0040 (16)
C12	0.043 (2)	0.0212 (19)	0.048 (2)	0.0007 (18)	-0.002 (2)	-0.0060 (17)
C13	0.048 (3)	0.029 (2)	0.060 (3)	-0.006 (2)	0.012 (2)	-0.001 (2)
C14	0.063 (3)	0.037 (2)	0.038 (2)	-0.003 (2)	0.009 (2)	0.0048 (19)

C15	0.041 (2)	0.029 (2)	0.037 (2)	0.0072 (18)	-0.0038 (19)	0.0001 (18)
C16	0.056 (3)	0.041 (2)	0.052 (2)	0.015 (2)	-0.009 (2)	-0.004 (2)
C17	0.045 (3)	0.042 (2)	0.048 (2)	0.005 (2)	-0.010 (2)	-0.006(2)
C18	0.053 (3)	0.028 (2)	0.039 (2)	0.0057 (19)	0.004 (2)	-0.0037 (18)
C19	0.038 (2)	0.034 (2)	0.041 (2)	-0.0032 (19)	0.0051 (19)	0.0015 (18)
C20	0.045 (3)	0.043 (3)	0.069 (3)	-0.004 (2)	-0.007 (2)	-0.014 (2)
C21	0.071 (3)	0.038 (2)	0.030 (2)	0.004 (2)	-0.009(2)	-0.0035 (18)
F1	0.288 (8)	0.198 (6)	0.107 (4)	-0.049 (6)	-0.005 (5)	-0.044 (5)
C22	0.055 (5)	0.014 (4)	0.008 (4)	-0.014 (4)	0.002 (3)	-0.005 (3)
C23	0.061 (6)	0.041 (5)	0.019 (4)	-0.018 (4)	-0.014 (4)	-0.004 (3)
C24	0.106 (8)	0.062 (6)	0.023 (4)	-0.015 (6)	-0.024 (5)	-0.021 (4)
C25	0.096 (10)	0.087 (8)	0.066 (7)	-0.007 (7)	0.029 (7)	-0.013 (6)
C26	0.120 (8)	0.080(7)	0.018 (5)	0.027 (6)	0.015 (5)	-0.020 (5)
C27	0.042 (6)	0.055 (6)	0.028 (4)	0.007 (5)	0.007 (4)	-0.012 (4)
F1'	0.198 (12)	0.197 (12)	0.197 (12)	-0.005 (9)	-0.010 (9)	-0.015 (9)
C22'	0.078 (8)	0.054 (7)	0.051 (6)	-0.022 (7)	-0.004 (6)	-0.004 (5)
C23'	0.075 (8)	0.079 (8)	0.089 (8)	-0.011 (7)	0.000(7)	0.016 (7)
C24'	0.099 (9)	0.071 (7)	0.029 (5)	-0.020(7)	-0.010 (6)	-0.018 (5)
C25'	0.116 (9)	0.090 (8)	0.073 (8)	-0.004 (7)	0.044 (7)	-0.001 (7)
C26'	0.176 (11)	0.091 (9)	0.070 (8)	-0.017 (8)	-0.047 (8)	0.001 (7)
C27'	0.091 (9)	0.094 (9)	0.065 (7)	-0.023 (7)	0.010 (7)	-0.014 (6)
C28	0.040 (3)	0.037 (2)	0.064 (3)	0.006 (2)	0.003 (2)	-0.001 (2)
C29	0.047 (3)	0.056 (3)	0.086 (4)	0.000(2)	0.022 (3)	0.001 (3)
C30	0.039 (2)	0.042 (2)	0.035 (2)	-0.002 (2)	-0.004 (2)	-0.0008 (19)
C31	0.047 (3)	0.057 (3)	0.037 (2)	-0.003 (2)	0.002 (2)	-0.003 (2)
C32	0.053 (3)	0.054 (3)	0.032 (2)	-0.003 (2)	0.003 (2)	-0.003 (2)
C33	0.059 (3)	0.035 (2)	0.038 (2)	0.009 (2)	0.009 (2)	-0.006 (2)
C34	0.085 (4)	0.043 (3)	0.054 (3)	0.012 (3)	0.033 (3)	0.003 (2)
C35	0.085 (4)	0.036 (3)	0.068 (3)	-0.006 (2)	0.043 (3)	-0.008 (2)
C36	0.063 (5)	0.275 (13)	0.162 (8)	-0.039 (6)	0.010 (6)	-0.044 (8)
C37	0.139 (8)	0.163 (8)	0.086 (5)	-0.033 (6)	0.034 (6)	-0.006 (5)
C38	0.127 (7)	0.123 (6)	0.161 (8)	-0.010 (6)	-0.071 (7)	-0.028 (6)
C39	0.081 (5)	0.173 (9)	0.187 (9)	-0.011 (6)	-0.012 (6)	-0.039(7)

### Geometric parameters (Å, °)

Cl1—C32	1.766 (4)	C14—H14A	0.9700
Cl2—C32	1.754 (4)	C14—H14B	0.9700
Cl3—C32	1.764 (4)	C15—C17	1.546 (5)
Cl4—C35	1.765 (6)	C15—C16	1.547 (5)
Cl5—C35	1.766 (4)	C16—H16A	0.9600
Cl6—C35	1.756 (5)	C16—H16B	0.9600
O1—C1	1.429 (4)	C16—H16C	0.9600
O1—H1	0.82 (2)	C17—H17A	0.9600
O2—C21	1.340 (5)	С17—Н17В	0.9600
O2—C2	1.452 (4)	С17—Н17С	0.9600
O3—C21	1.203 (5)	C18—H18A	0.9700
O4—C28	1.358 (5)	C18—H18B	0.9700
O4—C4	1.447 (4)	C19—H19A	0.9600

O5—C28	1.176 (5)	C19—H19B	0.9600
O6—C18	1.433 (4)	С19—Н19С	0.9600
O6—C5	1.444 (4)	C20—H20A	0.9600
O7—C30	1.323 (4)	С20—Н20В	0.9600
O7—C7	1.459 (4)	С20—Н20С	0.9600
O8—C30	1.187 (5)	C21—C22	1.425 (6)
O9—C30	1.337 (4)	C21—C22'	1.607 (9)
O9—C31	1.432 (4)	F1—C24	1.223 (8)
O10—C9	1.194 (4)	C22—C23	1.3900
O11—C33	1.337 (5)	C22—C27	1.3900
O11—C10	1.463 (4)	C23—C24	1.3900
O12—C33	1.182 (5)	С23—Н23	0.9300
O13—C33	1.344 (5)	C24—C25	1.3900
O13—C34	1.433 (5)	C25—C26	1.3900
O14—C13	1.419 (5)	С25—Н25	0.9300
O14—H14	0.83 (2)	C26—C27	1.3900
O15—C37	1.244 (8)	С26—Н26	0.9300
O16—C37	1.292 (9)	С27—Н27	0.9300
O16—C38	1.480 (8)	F1'—C24'	1.208 (9)
O17—H17D	0.91 (2)	C22'—C23'	1.3900
O17—H17E	0.89 (2)	C22'—C27'	1.3900
C1—C14	1.532 (6)	C23'—C24'	1.3900
C1—C15	1.556 (5)	C23'—H23'	0.9300
C1—C2	1.576 (5)	C24'—C25'	1.3900
С2—С3	1.559 (5)	C25'—C26'	1.3900
С2—Н2	0.9800	С25'—Н25'	0.9300
C3—C4	1.556 (5)	C26'—C27'	1.3900
C3—C8	1.594 (5)	C26'—H26'	0.9300
С3—Н3	0.9800	С27'—Н27'	0.9300
C4—C5	1.533 (5)	C28—C29	1.501 (6)
C4—C18	1.537 (5)	С29—Н29А	0.9600
C5—C6	1.520 (5)	С29—Н29В	0.9600
С5—Н5	0.9800	С29—Н29С	0.9600
C6—C7	1.523 (5)	C31—C32	1.499 (5)
С6—Н6А	0.9700	C31—H31A	0.9700
С6—Н6В	0.9700	C31—H31B	0.9700
С7—С8	1.550 (5)	C34—C35	1.516 (6)
С7—Н7	0.9800	C34—H34A	0.9700
C8—C19	1.525 (5)	C34—H34B	0.9700
C8—C9	1.562 (5)	C36—C37	1.444 (10)
C9—C10	1.526 (5)	С36—Н36А	0.9600
C10—C11	1.500 (5)	С36—Н36В	0.9600
С10—Н10	0.9800	С36—Н36С	0.9600
C11—C12	1.344 (5)	C38—C39	1.427 (9)
C11—C15	1.532 (5)	C38—H38A	0.9700
C12—C20	1.504 (5)	С38—Н38В	0.9700
C12—C13	1.504 (5)	С39—Н39А	0.9600
C13—C14	1.522 (5)	С39—Н39В	0.9600
C13—H13	0.9800	С39—Н39С	0.9600

C1—O1—H1	119 (4)	H19A—C19—H19B	109.5
C21—O2—C2	119.3 (3)	С8—С19—Н19С	109.5
C28—O4—C4	116.1 (3)	H19A—C19—H19C	109.5
C18—O6—C5	91.0 (2)	H19B—C19—H19C	109.5
C30—O7—C7	114.9 (3)	С12—С20—Н20А	109.5
C30—O9—C31	113.5 (3)	С12—С20—Н20В	109.5
C33—O11—C10	112.8 (3)	H20A—C20—H20B	109.5
C33—O13—C34	111.7 (3)	С12—С20—Н20С	109.5
C13—O14—H14	118 (3)	H20A—C20—H20C	109.5
C37—O16—C38	115.7 (6)	H20B-C20-H20C	109.5
H17D—O17—H17E	103 (3)	O3—C21—O2	124.5 (4)
O1—C1—C14	111.8 (3)	O3—C21—C22	117.7 (5)
O1—C1—C15	106.5 (3)	O2—C21—C22	117.7 (5)
C14—C1—C15	110.6 (3)	O3—C21—C22'	136.0 (6)
O1—C1—C2	103.7 (3)	O2—C21—C22'	99.4 (6)
C14—C1—C2	111.6 (3)	C22—C21—C22'	18.4 (4)
C15—C1—C2	112.4 (3)	C23—C22—C27	120.0
O2—C2—C3	108.0 (3)	C23—C22—C21	121.4 (4)
O2—C2—C1	103.5 (3)	C27—C22—C21	118.6 (4)
C3—C2—C1	118.6 (3)	C22—C23—C24	120.0
O2—C2—H2	108.8	С22—С23—Н23	120.0
С3—С2—Н2	108.8	C24—C23—H23	120.0
C1—C2—H2	108.8	F1—C24—C25	114.2 (7)
C4—C3—C2	112.4 (3)	F1—C24—C23	124.8 (7)
C4—C3—C8	110.9 (3)	C25—C24—C23	120.0
C2—C3—C8	115.3 (3)	C26—C25—C24	120.0
С4—С3—Н3	105.8	C26—C25—H25	120.0
С2—С3—Н3	105.8	C24—C25—H25	120.0
С8—С3—Н3	105.8	C27—C26—C25	120.0
O4—C4—C5	113.0 (3)	C27—C26—H26	120.0
O4—C4—C18	110.5 (3)	C25—C26—H26	120.0
C5—C4—C18	83.9 (3)	C26—C27—C22	120.0
O4—C4—C3	107.9 (3)	С26—С27—Н27	120.0
C5—C4—C3	120.5 (3)	С22—С27—Н27	120.0
C18—C4—C3	119.4 (3)	C23'—C22'—C27'	120.0
O6—C5—C6	113.6 (3)	C23'—C22'—C21	109.8 (7)
O6—C5—C4	92.0 (3)	C27'—C22'—C21	129.3 (7)
C6—C5—C4	119.9 (3)	C24'—C23'—C22'	120.0
O6—C5—H5	110.0	C24'—C23'—H23'	120.0
С6—С5—Н5	110.0	C22'—C23'—H23'	120.0
С4—С5—Н5	110.0	F1'—C24'—C23'	142.2 (8)
C5—C6—C7	112.6 (3)	F1'—C24'—C25'	97.5 (8)
С5—С6—Н6А	109.1	C23'—C24'—C25'	120.0
С7—С6—Н6А	109.1	C26'—C25'—C24'	120.0
С5—С6—Н6В	109.1	C26'—C25'—H25'	120.0
С7—С6—Н6В	109.1	C24'—C25'—H25'	120.0
Н6А—С6—Н6В	107.8	C27'—C26'—C25'	120.0
07—C7—C6	107.4 (3)	C27'—C26'—H26'	120.0
O7—C7—C8	107.9 (3)	C25'—C26'—H26'	120.0

C6—C7—C8	115.0 (3)	C26'—C27'—C22'	120.0
О7—С7—Н7	108.8	C26'—C27'—H27'	120.0
С6—С7—Н7	108.8	C22'—C27'—H27'	120.0
С8—С7—Н7	108.8	O5—C28—O4	124.2 (4)
C19—C8—C7	112.5 (3)	O5—C28—C29	125.1 (4)
C19—C8—C9	107.0 (3)	O4—C28—C29	110.7 (4)
С7—С8—С9	104.5 (3)	С28—С29—Н29А	109.5
C19—C8—C3	113.0 (3)	С28—С29—Н29В	109.5
C7—C8—C3	103.7 (3)	H29A—C29—H29B	109.5
C9—C8—C3	115.9 (3)	С28—С29—Н29С	109.5
O10—C9—C10	119.6 (3)	H29A—C29—H29C	109.5
O10—C9—C8	119.4 (3)	H29B—C29—H29C	109.5
C10—C9—C8	120.3 (3)	O8—C30—O7	127.4 (4)
O11—C10—C11	110.5 (3)	08—C30—O9	125.9 (4)
O11—C10—C9	108.6 (3)	07—C30—O9	106.7 (3)
C11—C10—C9	114.1 (3)	O9—C31—C32	108.2 (3)
O11—C10—H10	107.8	O9—C31—H31A	110.1
C11—C10—H10	107.8	C32—C31—H31A	110.1
C9—C10—H10	107.8	O9—C31—H31B	110.1
C12—C11—C10	119.5 (3)	C32—C31—H31B	110.1
C12—C11—C15	119.1 (3)	H31A—C31—H31B	108.4
C10—C11—C15	120.7 (3)	C31—C32—Cl2	107.2 (3)
C11—C12—C20	124.6 (4)	C31—C32—Cl3	110.9 (3)
C11—C12—C13	119.4 (4)	Cl2—C32—Cl3	110.0 (2)
C20—C12—C13	115.9 (3)	C31—C32—Cl1	110.3 (3)
O14—C13—C12	112.5 (4)	Cl2—C32—Cl1	109.7 (2)
O14—C13—C14	106.8 (3)	Cl3—C32—Cl1	108.8 (2)
C12—C13—C14	112.1 (3)	O12—C33—O11	127.7 (4)
O14—C13—H13	108.5	O12—C33—O13	125.2 (4)
С12—С13—Н13	108.5	O11—C33—O13	107.1 (4)
C14—C13—H13	108.5	O13—C34—C35	108.0 (4)
C13—C14—C1	118.1 (3)	O13—C34—H34A	110.1
C13—C14—H14A	107.8	С35—С34—Н34А	110.1
C1—C14—H14A	107.8	O13—C34—H34B	110.1
C13—C14—H14B	107.8	C35—C34—H34B	110.1
C1—C14—H14B	107.8	H34A—C34—H34B	108.4
H14A—C14—H14B	107.1	C34—C35—Cl6	107.5 (3)
C11—C15—C17	115.9 (3)	C34—C35—Cl4	112.0 (3)
C11—C15—C16	109.9 (3)	Cl6—C35—Cl4	109.0 (3)
C17—C15—C16	104.3 (3)	C34—C35—Cl5	110.1 (3)
C11—C15—C1	106.0 (3)	Cl6—C35—Cl5	109.3 (3)
C17—C15—C1	110.8 (3)	Cl4—C35—Cl5	108.9 (3)
C16—C15—C1	110.1 (3)	С37—С36—Н36А	109.5
C15—C16—H16A	109.5	С37—С36—Н36В	109.5
C15—C16—H16B	109.5	H36A—C36—H36B	109.5
H16A—C16—H16B	109.5	С37—С36—Н36С	109.5
C15—C16—H16C	109.5	H36A—C36—H36C	109.5
H16A—C16—H16C	109.5	H36B—C36—H36C	109.5
H16B—C16—H16C	109.5	O15—C37—O16	116.5 (9)

С15—С17—Н17А	109.5	O15—C37—C36	132.5 (9)
С15—С17—Н17В	109.5	O16—C37—C36	111.0 (7)
H17A—C17—H17B	109.5	C39—C38—O16	109.7 (6)
С15—С17—Н17С	109.5	С39—С38—Н38А	109.7
H17A—C17—H17C	109.5	O16—C38—H38A	109.7
H17B—C17—H17C	109.5	С39—С38—Н38В	109.7
O6—C18—C4	92.3 (3)	O16-C38-H38B	109.7
O6—C18—H18A	113.2	H38A—C38—H38B	108.2
C4C18H18A	113.2	С38—С39—Н39А	109.5
O6—C18—H18B	113.2	С38—С39—Н39В	109.5
C4C18H18B	113.2	H39A—C39—H39B	109.5
H18A—C18—H18B	110.6	С38—С39—Н39С	109.5
C8—C19—H19A	109.5	Н39А—С39—Н39С	109.5
С8—С19—Н19В	109.5	Н39В—С39—Н39С	109.5
C21—O2—C2—C3	128.3 (3)	C12—C11—C15—C17	175.8 (3)
C21—O2—C2—C1	-105.2 (3)	C10-C11-C15-C17	4.9 (5)
01—C1—C2—O2	60.6 (3)	C12-C11-C15-C16	-66.3 (4)
C14—C1—C2—O2	-59.9 (4)	C10-C11-C15-C16	122.8 (4)
C15—C1—C2—O2	175.2 (3)	C12—C11—C15—C1	52.6 (4)
O1—C1—C2—C3	-179.9 (3)	C10-C11-C15-C1	-118.3 (3)
C14—C1—C2—C3	59.6 (4)	O1-C1-C15-C11	-177.1 (3)
C15—C1—C2—C3	-65.3 (4)	C14—C1—C15—C11	-55.4 (4)
O2—C2—C3—C4	-11.9 (4)	C2-C1-C15-C11	70.0 (4)
C1—C2—C3—C4	-129.0 (3)	O1—C1—C15—C17	56.6 (4)
O2—C2—C3—C8	-140.3 (3)	C14—C1—C15—C17	178.2 (3)
C1—C2—C3—C8	102.6 (4)	C2-C1-C15-C17	-56.4 (4)
C28—O4—C4—C5	-56.2 (4)	O1-C1-C15-C16	-58.3 (4)
C28—O4—C4—C18	-148.2 (3)	C14—C1—C15—C16	63.4 (4)
C28—O4—C4—C3	79.6 (4)	C2-C1-C15-C16	-171.2 (3)
C2—C3—C4—O4	74.7 (4)	C5—O6—C18—C4	-7.2 (3)
C8—C3—C4—O4	-154.6 (3)	O4—C4—C18—O6	119.2 (3)
C2—C3—C4—C5	-153.4 (3)	C5—C4—C18—O6	6.9 (3)
C8—C3—C4—C5	-22.7 (4)	C3—C4—C18—O6	-114.8 (3)
C2—C3—C4—C18	-52.4 (4)	C2—O2—C21—O3	-4.2 (6)
C8—C3—C4—C18	78.3 (4)	C2—O2—C21—C22	179.4 (4)
C18—O6—C5—C6	131.3 (3)	C2—O2—C21—C22'	177.5 (5)
C18—O6—C5—C4	7.3 (3)	O3—C21—C22—C23	15.6 (7)
04—C4—C5—O6	-116.5 (3)	O2—C21—C22—C23	-167.8 (4)
C18—C4—C5—O6	-6.8 (3)	C22'—C21—C22—C23	-162 (3)
C3—C4—C5—O6	113.9 (3)	O3—C21—C22—C27	-165.1 (5)
O4—C4—C5—C6	124.7 (4)	O2—C21—C22—C27	11.5 (7)
C18—C4—C5—C6	-125.7 (4)	C22'—C21—C22—C27	18 (2)
C3—C4—C5—C6	-5.0 (5)	C27—C22—C23—C24	0.0
O6—C5—C6—C7	-111.6 (4)	C21—C22—C23—C24	179.2 (7)
C4—C5—C6—C7	-4.4 (5)	C22—C23—C24—F1	168.3 (10)
C30—O7—C7—C6	94.0 (4)	C22—C23—C24—C25	0.0
C30—O7—C7—C8	-141.5 (3)	F1-C24-C25-C26	-169.4 (9)
C5—C6—C7—O7	164.3 (3)	C23—C24—C25—C26	0.0
C5—C6—C7—C8	44.2 (4)	C24—C25—C26—C27	0.0

O7—C7—C8—C19	-68.0 (4)	C25—C26—C27—C22	0.0
C6—C7—C8—C19	51.8 (4)	C23—C22—C27—C26	0.0
O7—C7—C8—C9	47.7 (3)	C21—C22—C27—C26	-179.3 (7)
C6—C7—C8—C9	167.5 (3)	O3—C21—C22'—C23'	2.6 (11)
O7—C7—C8—C3	169.6 (3)	O2—C21—C22'—C23'	-179.5 (5)
C6—C7—C8—C3	-70.6 (4)	C22—C21—C22'—C23'	5.9 (19)
C4—C3—C8—C19	-66.2 (4)	O3—C21—C22'—C27'	-166.8 (7)
C2—C3—C8—C19	62.9 (4)	O2—C21—C22'—C27'	11.2 (9)
C4—C3—C8—C7	55.9 (3)	C22—C21—C22'—C27'	-163 (3)
C2—C3—C8—C7	-175.0 (3)	C27'—C22'—C23'—C24'	0.0
C4—C3—C8—C9	169.8 (3)	C21—C22'—C23'—C24'	-170.5 (9)
C2—C3—C8—C9	-61.1 (4)	C22'—C23'—C24'—F1'	172 (2)
C19—C8—C9—O10	18.4 (4)	C22'—C23'—C24'—C25'	0.0
C7—C8—C9—O10	-101.1 (4)	F1'-C24'-C25'-C26'	-175.2 (13)
C3—C8—C9—O10	145.4 (3)	C23'—C24'—C25'—C26'	0.0
C19—C8—C9—C10	-170.8 (3)	C24'—C25'—C26'—C27'	0.0
C7—C8—C9—C10	69.7 (4)	C25'—C26'—C27'—C22'	0.0
C3—C8—C9—C10	-43.8 (4)	C23'—C22'—C27'—C26'	0.0
C33—O11—C10—C11	-158.6 (3)	C21—C22'—C27'—C26'	168.4 (11)
C33—O11—C10—C9	75.6 (4)	C4—O4—C28—O5	-2.3 (6)
O10—C9—C10—O11	-5.4 (5)	C4—O4—C28—C29	178.2 (3)
C8—C9—C10—O11	-176.2 (3)	C7—O7—C30—O8	2.3 (6)
O10-C9-C10-C11	-129.2 (4)	C7—O7—C30—O9	-177.5 (3)
C8—C9—C10—C11	60.0 (4)	C31—O9—C30—O8	9.1 (6)
O11—C10—C11—C12	115.7 (4)	C31—O9—C30—O7	-171.1 (3)
C9-C10-C11-C12	-121.6 (4)	C30—O9—C31—C32	-174.4 (3)
O11—C10—C11—C15	-73.5 (4)	O9—C31—C32—Cl2	180.0 (3)
C9—C10—C11—C15	49.2 (4)	O9—C31—C32—Cl3	59.9 (4)
C10-C11-C12-C20	-11.8 (5)	O9—C31—C32—C11	-60.7 (4)
C15—C11—C12—C20	177.3 (3)	C10—O11—C33—O12	9.0 (6)
C10-C11-C12-C13	165.0 (3)	C10—O11—C33—O13	-171.1 (3)
C15—C11—C12—C13	-6.0 (5)	C34—O13—C33—O12	0.0 (6)
C11—C12—C13—O14	-155.7 (3)	C34—O13—C33—O11	-179.8 (3)
C20-C12-C13-O14	21.3 (5)	C33—O13—C34—C35	170.1 (4)
C11—C12—C13—C14	-35.3 (5)	O13—C34—C35—Cl6	178.6 (3)
C20-C12-C13-C14	141.6 (4)	O13—C34—C35—Cl4	58.9 (4)
O14-C13-C14-C1	151.0 (3)	O13—C34—C35—Cl5	-62.5 (5)
C12-C13-C14-C1	27.4 (5)	C38—O16—C37—O15	3.0 (11)
O1-C1-C14-C13	136.6 (4)	C38—O16—C37—C36	-177.4 (8)
C15-C1-C14-C13	18.1 (5)	C37—O16—C38—C39	171.4 (8)
C2-C1-C14-C13	-107.8 (4)		

### Hydrogen-bond geometry (Å, °)

D—H···A	<i>D</i> —Н	H···A	$D \cdots A$	$D\!\!-\!\!\mathrm{H}^{\dots}\!A$
O17—H17E···Cl1 <sup>i</sup>	0.89 (2)	2.85 (6)	3.582 (5)	141 (6)
017—H17E···O9 <sup>i</sup>	0.89 (2)	2.48 (7)	3.241 (6)	143 (9)
O14—H14…O17	0.83 (2)	2.00 (2)	2.791 (6)	161 (4)
O1—H1…O15	0.82 (2)	2.01 (3)	2.786 (6)	158 (5)

O17—H17D····O1 <sup>ii</sup>	0.91 (2)	2.25 (4)	3.085 (6)	152 (6)
Symmetry codes: (i) $-x+1$ , $y+1/2$ , $-z+3/2$ ; (ii) $x+1/2$ ,	-y+1/2, -z+1.			





Fig. 2

